

Europäisches Patentamt

European Patent Office

Office européen des brevets



(11) EP 0 895 013 A3

(12)

EUROPEAN PATENT APPLICATION

(88) Date of publication A3: 31.05.2000 Bulletin 2000/22

(51) Int. Cl.⁷: **F16L 11/12**, F16L 11/08

(43) Date of publication A2: 03.02.1999 Bulletin 1999/05

(21) Application number: 98113796.1

(22) Date of filing: 23.07.1998

(84) Designated Contracting States:

AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT SE
Designated Extension States:
AL LT LV MK RO SI

(30) Priority: 29.07.1997 US 901972

(71) Applicant: DAYCO PRODUCTS, INC. Miamisburg, OH 45342 (US)

(72) Inventors:

Little, Kent H.
 Ozark, Michigan 65721 (US)

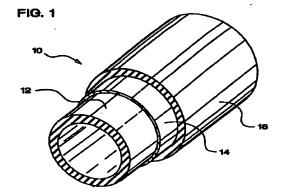
Barnes, John A.
 Springfield, Missouri 65802 (US)

(74) Representative:

Splanemann Reitzner Baronetzky Westendorp Patentanwälte Rumfordstrasse 7 80469 München (DE)

(54) Oxygen barrier structure for a flexible hose

(57) A flexible hose (10, 20) for use in heating systems has an aluminum barrier layer (14, 24) for preventing ingestion of oxygen. The aluminum barrier layer (14, 24) is securely bonded between two adhesion tubes (12, 22; 16, 26) which are vulcanized in place against the aluminum. Vulcanization is performed in the presence of peroxide. Principal active ingredients of the adhesion tubes (12, 22; 16, 26) are ethylene propylene diene polymethylene and polybutadiene-maleic anhydride adduct resin.



EP 0 895 013 A3



EUROPEAN SEARCH REPORT

Application Number EP 98 11 3796

-		DERED TO BE RELEVA	<u> </u>	
Category	Citation of document with of relevant pa	n indication, where appropriate, ssages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
A	US 5 526 848 A (TE AL) 18 June 1996 (* column 1, line 3 * column 3 - colum	9 - line 52 *	1-3,6,7,	F16L11/12 F16L11/08
Α	/ December 1988 (1	KOHAMA RUBBER CO LTD) 988-12-07) - page 3, line 8; fig	9	
	US 5 647 400 A (JA 15 July 1997 (1997 * column 3 * * column 6, line 6		1-3,9	
İ	6 February 1996 (1	ILES DANIEL T ET AL) 996-02-06) - column 4, line 9;	1,2,9	
	WO 93 25835 A (ITT 23 December 1993 (* page 32, line 25	1993–12–23)	1	TECHNICAL FIELDS SEARCHED (Int.CI.6)
İ				
	The present search report has	heen drawn up for all claims		
	Place of search	Date of completion of the searc		
	BERLIN	5 April 2000	i	Examiner
CAT X : particu Y : particu docum	TEGORY OF CITED DOCUMENTS Ilarly relevant if taken alone Ilarly relevant if combined with anole ient of the same category logical background	T : theory or print E : earlier pater after the filin her D : document ci	nciple underlying the innt document, but publish g date g date ted in the application ted for other reasons	ned on, or
O:non-w	vritten disclosure ediate document	& : member of t document	he same patent family,	corresponding

EPO FORM 1503 03.82 (P04C01)

ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 98 11 3796

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

05-04-2000

Patent document cited in search report			Publication date		Patent family member(s)	Publication date
US	5526848	Α	18-06-1996	JP DE	6241354 A 4404099 A	30-08-199 11-08-199
EP	0294181	Α	07-12-1988	JP	1987241 C	08-11-199
				JP	6096280 B	30-11-199
				JP	63302036 A	08-12-198
				JP	2027310 C	26-02-199
				JP	5052781 B	06-08-199
				JP	63302037 A	08-12-198
				DE	3887386 D	10-03-199
				DE	3887386 T	26-05-199
				KR US	9606172 B 4905736 A	09-05-199 06-03-199
	5647400	Α	 15-07-1997	NONE		
				NUNE		
US	5488975	Α	06-02-1996	NONE		
WO	9325835	Α	23-12-1993	US	5383087 A	17-01-199
				DE	9321565 U	02-12-199
				DE	69326766 D	18-11-199
				DE	69326766 T	02-03-200
				EP	0644990 A	29-03-199
				JP	2812802 B	22-10-199
				JP	7507739 T	31-08-199
				US	5524673 A	11-06-199
				US US	5743304 A 5996642 A	28-04-199 07-12-199

FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

THIS PAGE BLANK (USPTO)

(11) EP 0 895 013 A2

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:

03.02.1999 Bulletin 1999/05

(51) Int. Cl.⁶: **F16L 11/12**, F16L 11/08

(21) Application number: 98113796.1

(22) Date of filing: 23.07.1998

(84) Designated Contracting States:

AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT SE

Designated Extension States: AL LT LV MK RO SI

(30) Priority: 29.07.1997 US 901972

(71) Applicant: DAYCO PRODUCTS, INC. Miamisburg, OH 45342 (US)

(72) Inventors:

Little, Kent H.
 Ozark, Michigan 65721 (US)

Barnes, John A.
 Springfield, Missouri 65802 (US)

(74) Representative:

Patentanwälte

Dipl.-Ing. R. Splanemann

Dr. B. Reitzner

Dipl.-Ing. K. Baronetzky

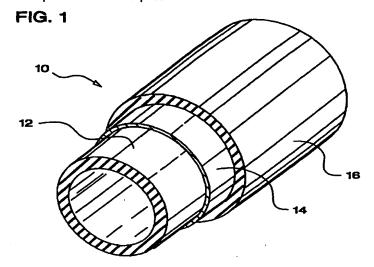
Tal 13

80331 München (DE)

(54) Oxygen barrier structure for a flexible hose

(57) A flexible hose (10, 20) for use in heating systems has an aluminum barrier layer (14, 24) for preventing ingestion of oxygen. The aluminum barrier layer (14, 24) is securely bonded between two adhesion tubes (12, 22; 16, 26) which are vulcanized in place against the aluminum. Vulcanization is performed in the pres-

ence of peroxide. Principal active ingredients of the adhesion tubes (12, 22; 16, 26) are ethylene propylene diene polymethylene and polybutadiene-maleic anhydride adduct resin.



EP 0 895 013 A2

Description

Background of the Invention

[0001] This invention relates to the field of flexible hoses and more particularly to flexible hoses used for circulation of fluids in radiant heating systems in homes and businesses. Typical prior art hoses for such applications are disclosed, for example, in Chiles et al. U.S. Patent No. 4,779,673 and in Chiles et al. U.S. Patent No. 5,488,975.

[0002] Often times such hoses supply water or other heating fluid to heat exchangers located in floors, ceilings, roofs and concrete or asphalt slabs. The hoses may be embedded in the surfaces to be heated, and it is desirable that they be flexible for ease of installation. A significant problem with such hoses is that they are subject to gas infiltration and exfiltration. Oxygen is particularly troublesome, because it is easily ingested from the outside environment. Once oxygen has gained entry to such a heating system, it deteriorates the hoses and corrodes the pumping system.

[0003] Chiles '975 discloses a flexible heating system hose having an oxygen barrier layer in the form of a thin plastic film such as ethylene vinyl alcohol. Alternatively, the patent suggests the use of an aluminum barrier layer. Both of these barriers have problems. Oxygen is able to penetrate all known plastic films, at least to some small degree. Aluminum is substantially impermeable to oxygen, but it does not bond well against other materials. Consequently prior art hoses having aluminum barrier layers tend to delaminate under stress.

[0004] A number of other prior art patents disclose flexible hoses incorporating metallic layers of one type or another. Such disclosures appear, for example, in Hane et al. U.S. Patent No. 4,559,973, Campbell et al. U.S. Patent No. 4,758,455, Davis U.S. Patent No. 5,182,147, Yoshikawa et al. U.S. Patent No. 5,271,977 and Spurgett U.S. Patent No. 5,398,729. None of these prior art references teach a flexible hose having an aluminum barrier bonded therein with sufficient adhesion to avoid delamination over a long period of time in a heating system environment. Typically such prior art hoses are used for fuel and vapor transmission and are constructed by coating the metal strip on both sides with an adhesive which may, for example, be an adhesive made from a copolymer of ethylene and a monomer having a reactive carboxyl group, as taught in Campbell et al.

[0005] It is therefore seen that there is a need for a flexible hose incorporating an aluminum barrier layer, and able to resist delamination in a heating system environment.

Summary of the Invention

[0006] This invention provides a flexible hose having an aluminum barrier layer sandwiched between two vulcanized elastomeric adhesion tubes. These adhesion tubes are securely bonded to the aluminum barrier layer by heat and pressure during assembly of the hose. The arrangement resists delamination and is substantially impermeable to oxygen. [0007] The adhesion tubes comprise at least about 20 percent by weight of ethylene propylene diene polymethylene and from about 2 to 9 percent by weight of polybutadiene-maleic anhydride adduct resin. The formulation also includes about 2-8 percent by weight of active peroxide as a critical curing agent. The peroxide cure is performed in a conventional manner at a temperature of about 325°F. The cure unexpectedly produces a strong bond between the aluminum and the adhesion tubes. Only a peroxide cure has been found to produce such results. Suitable conventional black and non-black filler ingredients, and paraffinic or naphthenic plasticizers may be added to the mixture, as desired.

[0008] It is therefore an object of the invention to provide an improved flexible hose for heating system applications.

[0009] It is another object of the invention to provide a flexible hose having a securely bonded oxygen barrier layer.

[0010] Other objects and advantages of the invention will be apparent from the following description, the accompanying drawings and the appended claims.

45 Brief Description of the Drawings

[0011]

50

30

Fig. 1 is a perspective view of the invention in its simplest form comprising only an aluminum barrier layer sandwiched between two adhesion tubes;

Fig. 2 is a plan view of a cut away plan view of a heating system hose embodying the present invertical;

<u>Detailed Description of the Preferred Embodiment</u>

[0012] A simple flexible hose made in accordance with this invention is illustrated in Fig. 1 as indicated by the reference numeral 10. Hose 10 comprises an aluminum barrier layer 14 sandwiched between an interiorly disposed first adhesion tube 12 and an exteriorly disposed second adhesion tube 16. Adhesion tubes 12,16 are formed from a plastic mass which is vulcanized in place against barrier layer 14.

[0013] The plastic mass is formulated from a mixture comprising at least about 20 percent by weight of ethylene propylene diene polymethylene (EPDM) and from about 2 to 9 percent by weight of polybutadiene-maleic anhydride adduct resin. These ranges are based upon adhesion testing of structures formulated from mixtures having as low as 29% by weight of EPDM, polybutadiene-maleic anhydride adduct resin in amounts as low as 4% by weight and as high as 7.3% by weight and peroxide in amounts as low as 1.3% by weight and as high as 3.9% by weight.

[0014] The mixture should incorporate from about 7 to 35 percent by weight of paraffinic oil, naphthenic hydrocarbon or other suitable plasticizer, and preferably also a copolymer of polyolefin elastomer in an amount less than about 7 percent by weight, as well as chlorosulfanated polyethylene in an amount less than about 4% by weight. These ranges have been verified by adhesion tests on structures formulated from mixtures comprising paraffinic oil in amounts as low as 13% by weight and as high as 18% by weight, a copolymer of polyolefin elastomer in an amount as high as 4 percent by weight, and chlorosulfanated polyethylene in an amount as high as 2% by weight. Finally, as is well known, 1 to 2% by weight of zinc oxide and up to 1.5 percent by weight of stearic acid should be added for satisfactory vulcanization.

[0015] Filler materials may be added to the mixture, as desired, to obtain well known rubber tube properties, typically in a range from about 17 to 80% by weight. Satisfactory adhesion has been demonstrated for tubes formulated from mixtures having a filler content as low as 36% and as high as 43%.

[0016] Table I presents the preferred formulation for a compound which may be converted into adhesion tubes such as tubes 12, 16. The preferred formulation uses a combination of low diene hexa EPDM and normal diene hexa EPDM. These ingredients are combined with polyolefin ethylene-octene copolymer and chlorosulfarated polyethylene, which are collectively referred to as "rubber". The total composition comprises 100 parts rubber. The table presents the specific gravity of each ingredient, the percentage by weight and also the number of parts per hundred parts rubber (PHR)

TABLE I

25					
	Ingredient	SpGr.	PHR	Wt.%	
30	low diene hexa EPDM	0.87	70.00	24.10	
	hexa EPDM	0.88	15.00	5.16	
	polyolefin ethylene-octene copolymer	0.91	10.00	3.44	
35	chlorosulfanated polyethylene	1.17	5.00	1.72	
	carbon black	1.80	80.00	27.54	
40	calcium carbonate	2.71	40.00	13.77	
	paraffinic oil	0.89	45.00	15.49	
	zinc oxide	5.60	4.00	1.38	
45	stearic acid	0.84	1.54	0.52	•
	50% DBPH peroxide solution	1.50	7.00	2.41	
	polybutadiene-maleic anhydride adduct resin	1.21	13.00	4.48	

50

[0017] The above ingredients are mixed at a temperature in a range between about 210 ° F and 250.° F which is high enough to promote plastic flow but too low for curing. The resulting plastic mass is extruded to produce a first adhesion tube such as tube 12. Tube 12 is then wrapped by a layer of aluminum foil which may be between about 0.0005 and 0.003 inches thick. This may be accomplished by helical wrapping or by tensioned radial curling. That produces the barrier layer 14 which is not yet bonded to tube 12.

[0018] After tube 12 has been extruded and wrapped with aluminum foil, a second adhesion tube 16 is formed about

and in surface-to-surface contact with barrier layer 14. Tube 16 is fabricated from a plastic mass as described above, and may be produced by a single step extrusion process. Alternatively, tube 16 may be formed by extruding the plastic mass into a sheet or a strip and then wrapping the sheet or strip about barrier layer 14. This produces the assembly 10 of Fig. 1. Assembly 10 is then cured in an autoclave for about 35 minutes at a temperature in a range between about 320 ° F and 330 ° F. This results in a surprisingly good bond between elastomeric tubes 12 and 16 and the barrier layer 14 which is sandwiched therebetween. It has been confirmed that the adhesion is related to the use of peroxide in the cure.

[0019] A series of tests have shown that good bonds cannot be obtained when sulfur is used as the curing agent. Whereas structures made in accordance with this invention were found in some cases to withstand delamination stresses in excess of . 68 p.p.i., similar structures, cured in the presence of sulfur, generally delaminated at stresses of less than 1 p.p.i. . Typical test results for structures cured in the presence of peroxide are presented in Table III. Table III identifies the ingredients referred to

[0020] in Table III.

15

40

45

50

55

TABLE II

20	Code	Ingredient
	Α	low diene hexa EPDM
	В	polyolefin ethylene-octene copolymer
25	С	hexa EPDM
	D	chlorosulfanated polyethylene
	E	carbon black
30	F	calcium carbonate
	G	paraffinic oil
35	Н	zinc oxide

	I	stearic acid
5	J `	A.A bis T. Butylperoxy 40% solution
	K	dicumyl peroxide, 60% solution
	L	DBPH peroxide 50% solution
10	M	polybutadiene-maleic anhydride adduct resin, 70% DLC

15

TABLE III

20	Ingredient	Amount (PHR)								
	Α	55.0	55.0	40.0	55.0	70.0	70.0	70.0	40.0	40.0
25	В	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0
	С	30.0	30.0	45.0	30.0	15.0	15.0	15.0	45.0	45.0
	D	5.0	5.0	5.0	5.0	5.0	5.0	5.0	5.0	5.0
30	E	· 80.0	75.0	70.0	70.0	75.0	80.0	70.0	75.0	80.0
	F	40.0	38.0	36.0	36.0	38.0	40.0	36.0	38.0	40.0
	G	44.0	42.0	40.0	40.0	42.0	44.0	40.0	42.0	44.0
35	Н	4.0	4.0	4.0	4.0	4.0	4.0	4.0	4.0	4.0
	I	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5
40	J	10.0	0.0	7.5	0.0	5.0	0.0	0.0	0.0	0.0
	K	0.0	0.0	0.0	3.3	0.0	5.0	0.0	6.7	0.0
45	L	0.0	6.5	0.0	0.0	0.0	0.0	8.7	0.0	4.35
	М	20.0	20.0	20.0	20.0	20.0	20.0	20.0	20.0	20.0
	Adhesion p.s.i	50.0	29.0	60.0	29.0	34.0	45.0	56.0	68.2	2.8

⁵⁰

[0022] Reinforcement layer 28 and cover 30 may be of conventional construction. Thus reinforcement layer 28 may be fabricated from rayon, polyester, polyvinyl acetate, wire, aramid or other suitable material. Cover 30 may be pro-

^[0021] Fig. 2 illustrates a flexible heating system hose 20, made in accordance with this invention. The hose comprises an interiorly disposed first adhesion tube 22, an aluminum barrier layer 24 bonded exteriorly against first adhesion tube 22, and a second adhesion tube 26 bonded exteriorly against aluminum barrier layer 24. Tubes 22, 26 and barrier layer 24 correspond to tubes 12, 16 and barrier layer 14 of Fig. 1 and are fabricated in a similar manner. Hose 20 also has a reinforcement layer 28 bonded exteriorly against adhesion tube 26, and a cover 30 adhesively secured exteriorly against reinforcement layer 28.

duced from any of numerous thermosetting elastomeric compounds such as natural rubber, styrene butadiene, polychloroprene, acrylonitrile butadiene, chlorosulfanated monomer, chloronated polyethylene, ethylene propylene monomer or ethylene propylene diene polymethylene.

[0023] While the forms of apparatus herein described and the method of making them constitute preferred embodiments of this invention, it is to be understood that the invention is not limited to these precise forms of apparatus or method, and that changes may be made therein without departing from the scope of the invention which is defined in the appended claims.

Claims

10

15

- 1. A hose comprising a first tube (12, 22) and a second tube (16, 26) disposed about said first tube, characterized in that an aluminum foil (14, 24) is bonded exteriorly about said first tube (12, 22), and said second tube (16, 26) is bonded exteriorly about said aluminum foil (14, 24); said first tube and said second tube each comprising at least about 20 percent by weight of ethylene propylene diene polymethylene and from about 2 to 9 percent by weight of polybutadiene-maleic anhydride adduct resin, said first tube (12, 22) and said second tube (16, 26) having been bonded to said aluminum foil (14, 24) by thermal curing in the presence of peroxide in the amount of 0.6 percent to 2.8 percent by weight of tube material.
- 2. A hose according to claim 1 further characterized by a reinforcement layer (28) bonded exteriorly about said second tube (26), and a cover (30) bonded exteriorly about said reinforcement layer (28).
 - A hose according to claim 2 characterized in that said first tube (12, 22) and said second tube (16, 26) each comprise about 7 percent to 35 percent by weight of paraffinic oil.
- 4. A hose according to claim 3 characterized in that said first tube (12, 22) and said second tube (16, 26) each comprise a copolymer of polyolefln elastomer in an amount less than about 7 percent by weight.
 - 5. A hose according to claim 4 characterized in that said first tube (12, 22) and said second tube (16, 26) each comprise chlorosulfanated polyethylene in an amount less than about 4 percent by weight.
- *30* **6.** A ho
 - A hose according to claim 5 characterized in that said first tube (12, 22) and said second tube (16, 26) each ocmprise about 1 percent to 2 percent by weight of zinc oxide.
- 7. A hose according to claim 6 characterized in that said first tube (12, 22) and said second tube (16, 26) each comprise stearic acid in an amount of less than about 1.5 percent by weight.
 - 8. A hose according to claim 7 characterized in that said first tube (12, 22) and said second tube (16, 26) each comprise about 17 percent to 80 percent by weight of filler material.
- 40 9. A thermally curable hose tube which is curably adherent to an aluminum surface and comprises at least about 20 percent by weight of ethylene propylene diene polymethylene, about 2 to 9 percent by weight of polybutadiene-maleic anhydride adduct resin and about 0.6 to 2.8 percent by weight of peroxide, and perferably further comprises about 7 percent to 35 percent by weight of paraffinic oil, a copolymer of polyoelfin elastomer in an amount less than about 7 percent by weight, chlorosulfanted polyethylene in an amount of less than about 4 percent by weight, and about 1 percent to 2 percent by weight of zinc oxide, stearic acid in an amount of less than about 1.5 percent by weight, and about 17 percent to 80 percent by weight of filler material.
 - 10. A method of making a flexiable hose characterized by the steps of:
- 50

55

- (1) preparing a mixture comprising at least about 20 percent by weight of ethylene propylene diene polymethylene, about 2 to 9 percent by weight of polybutadiene-maleic anhydride adduct resin, about 0.6 to 2.8 percent by weight of peroxide, about 7 percent to 35 percent by weight of paraffinic oil, a copolymer of polyolefin elastomer in an amount less than about 7 percent by weight, chlorosulfanated polyethylene in an amount of less than about 4 percent by weight, about 1 to 2 percent by weight of zinc oxide, stearic acid in an amount of less than about 1.5 percent by weight, and about 17 percent to 80 percent by weight of filler material;
- (2) creating a plastic mass by mixing said mixture at a temperature high enough to obtain a plastic flow but below the curing temperature thereof;
- (3) forming a first tube (12, 22) from said plastic mass;

- (4) wrapping a layer of aluminum foil (14, 24) about said first tube (12, 22) and in surface contact therewith;
- (5) forming a second tube (16, 26) from said plastic mass, said second tube being formed exteriorly about said layer of aluminum foil (14, 24) and in surface contact therewith; and
- (6) thermally curing said first tube (12, 22) and said second tube (16, 26) while maintaining surface contact thereof against said layer of aluminum foil (14, 24).

